

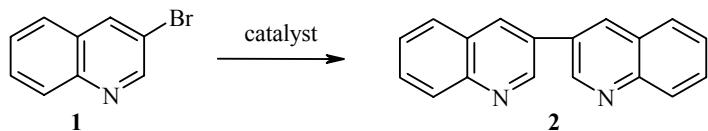
HETEROGENOUS CATALYTIC METHOD FOR THE SYNTHESIS OF BIQUINOLINES AND BIPYRIDINES

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A method has been developed for the synthesis of 2,2'- and 3,3'-biquinolines and of 4,4'-bipyridine based on the coupling of 2- and 3-bromoquinoline or 4-bromopyridine using a Pd/C-hydrazine-KOH catalytic system.

Keywords: 2,2'-biquinoline, 3,3'-biquinoline, 4,4'-bipyridine, 4-bromopyridine, 2-bromoquinoline, 3-bromoquinoline, palladium on carbon, coupling.

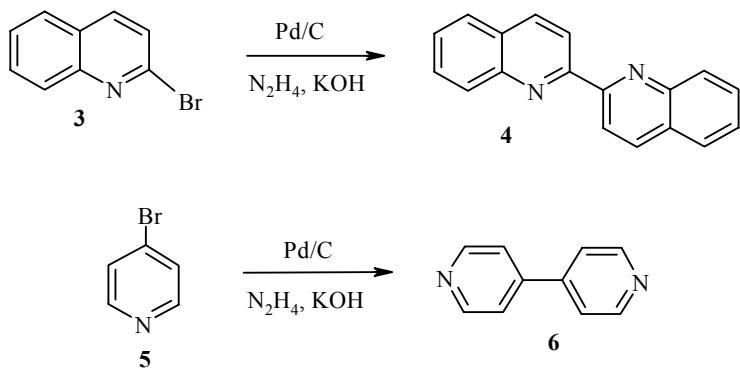
A series of methods for the synthesis of 3,3'-biquinoline (**2**) is known [1-4]. These methods are based on the cross coupling of 3-bromoquinoline (**1**) under the action of different metals but have a number of deficiencies. The method [1] using equimolar amounts of 3-bromoquinoline and catalyst employs the expensive metallic palladium. Despite the fact that in method [2] a catalytic amount of metallic palladium is involved it cannot be used repeatedly. The deficiencies of methods [3] and [4] using a catalytic system based on Ni(0) are the need for chromatographic separation of the reaction mixture which impedes work up of large quantities of the 3,3'-biquinoline. Hence we decided to develop a more efficient synthesis of compound **2**.



We have attempted to develop a method allowing a repeated use of the palladium catalyst based on the work in [1]. It was found that 10% Pd/C can be used in place of palladium black. It can be regenerated using the system hydrazine–KOH. The yield is virtually unchanged at 46%.

It was further shown that the catalyst can be regenerated in the course of the reaction by gradual addition to the reaction mixture of a solution of alkali in hydrazine hydrate. This permits use of a catalytic amount of palladium and repeated use of the catalyst. However, the yield in this case decreases to 28%.

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This method can be used in the synthesis of other biquinolines and bipyridines, e.g. 2,2'-biquinoline (**4**) ad 4,4'-bipyridine (**6**).

EXPERIMENTAL

^1H NMR spectra were recorded on a Bruker WP-200 (200 MHz) instrument using TMS as internal standard. Monitoring of the reaction course and purity of the synthesized compounds was carried out on Silufol UV-254 plates with ethyl acetate as solvent.

General Method. A solution of KOH (0.67 g, 12 mmol) and 88% hydrazine hydrate (1.14 g, 20 mmol) in ethyl alcohol (3 ml) was added dropwise over 2 h to a mixture of the corresponding halo derivative (10 mmol) and Pd/C (0.51 g, 0.5 mmol) in refluxing ethanol (10 ml). Solvent was evaporated and the residue was recrystallized from DMF, separating the catalyst and KBr on a plaited filter. The catalyst was washed with water, dried, and used repeatedly.

3,3'-Biquinoline (2). Yield 0.72 g (28%) with mp 269-271°C (alcohol) (mp 217°C [1]). ^1H NMR spectrum, δ , ppm (J , Hz): 7.63 (2H, ddd, $J_{5,6} = 8.1$, $J_{6,7} = 6.9$, $J_{6,8} = 1.2$, H-6,6'); 7.77 (2H, ddd, $J_{6,7} = 6.9$, $J_{7,8} = 8.4$, $J_{5,7} = 1.5$, H-7,7'); 7.94 (2H, dd, $J_{5,6} = 8.1$, $J_{5,7} = 1.5$, H-5,5'); 8.18 (2H, dd, $J_{7,8} = 8.4$, $J_{6,8} = 1.2$, H-8,8'); 8.46 (2H, d, $J_{2,4} = 2.4$, H-4,4'); 9.29 (2H, d, $J_{2,4} = 2.4$, H-2,2'). Found, %: C 84.71; H 4.58; N 10.71. $\text{C}_{18}\text{H}_{12}\text{N}_2$. Calculated, %: C 84.37; H 4.69; N 10.94.

2,2'-Biquinoline (4, $\text{C}_{18}\text{H}_{12}\text{N}_2$). Yield 1 g (39%), mp 197-198°C (alcohol) (mp 196-198°C [5]). A sample mixed with a known sample (commercially available) did not give a melting point depression. The ^1H NMR spectrum was identical to that given in [5].

4,4'-Bipyridine (6, $\text{C}_{10}\text{H}_8\text{N}_2$). Yield 0.53 g (34%) with mp 112-114°C (water) (mp 112-114°C [5]). A sample mixed with a known sample (commercially available) did not give a melting point depression. The ^1H NMR spectrum was identical to that given in [5].

REFERENCES

1. K. Ueda, *Yakugaku Zasshi*, **51**, 495 (1931).
2. J. Hassan, V. Penalva, L. Laveton, Ch. Gozzi, and M. Lemaire, *Tetrahedron*, **54**, 13793 (1998).
3. Yv. Fort, S. Becker, and P. Caubère, *Tetrahedron*, **50**, 11893 (1994).
4. M. Iyoda, H. Otsuka, K. Sato, N. Nisato, and M. Oda, *Bull. Chem. Soc. Jpn.*, **63**, 80 (1990).
5. M. Tieco, L. Testaferri, M. Tingoli, D. Chianelli, and M. Montanucci, *Synthesis*, 736 (1984).